Conductivity of macromolecular networks measured by electrostatic force microscopy

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High-resolution electrostatic force microscopy (EFM)-phase measurements are reported on molecular networks of semiconductor polymer poly-hexylthiophene (P3HT) and DNA molecules. A lateral resolution of better than 20 nm is demonstrated in EFM-phase images of the P3HT network by detecting the phase shift of the tip along the molecules under electrical bias. Strands of A-DNA are shown to be highly insulating in comparison to the semiconductor polymer P3HT, with a minimum resistance of \( \sim 1 \times 10^7 \) \( \Omega \) cm. © 2003 American Institute of Physics.

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There is a continual search for a material systems and fabrication technologies for electrical discrete devices and complex networks, aiming at the smallest possible length scales. Semiconducting conjugated polymers have made rapid progress over recent years for display applications and for electronic devices. For instance, poly-hexylthiophene (P3HT) has been used to make field-effect transistors (FET) devices, and scanning Kelvin probe microscopy has been used to characterize them.1 Intense recent effort has focused on exploiting the inherent one-dimensional nature of carbon nanotubes (CNTs) and DNA molecules for devices on the nanometer length scale. FETs have been successfully produced with CNTs,2 DNA offers the capability to self-assemble in predetermined structures from strands having engineered base sequences. However, there has been considerable controversy regarding its conductivity; conducting, semiconducting, and insulating behavior have all been recently reported,3–5 suggesting that factors such as humidity and adsorbed water might play a significant role. The conductivity of DNA can be modified by doping of the bases,6 and adsorbed water might play a significant role in terms of local concentration gradients during growth. We perform electrical measurements of the network with the EFM-phase technique and use it to characterize modifications of the network produced by AFM manipulation of the P3HT molecules. We also apply the measurements to measure the electrical characteristics of partially-aligned \( \lambda \)-DNA strands, and find insulating behavior.

During an EFM measurement, the main scan records the surface topographical data in tapping mode. In the subsequent interleave scan, the tip is lifted and kept at fixed height (typically 20 nm) above the surface, whose topography is obtained in the first scan. During the interleave scan, the tip is kept driven in oscillation with the same driving frequency as in the main scan. The frequency (\( \omega \)) or phase (\( \phi \)) shift is a function of the surface potential, and are given by

\[
\Delta \omega = \frac{\omega_0}{2k} \frac{dF(z)}{dz}, \quad \Delta \phi = \frac{Q}{k} \frac{dF(z)}{dz},
\]

where \( F(z) \) is the electrostatic force between the tip and surface at the relative separation \( z \), \( Q \) is the cantilever quality factor, \( k \) is the cantilever spring constant. \( F(z) \) is related to the local capacitance (\( C \)) between the tip and the sample by

\[
F(z) = \frac{1}{2} \frac{dC}{dz} (\Delta U)^2,
\]

where \( \Delta U \) is the potential difference between the tip and the sample. In the conventional surface potential measurement or Kelvin probe method,9 the tip bias is varied to give \( F(z) = 0 \), and hence \( \Delta U = 0 \), so that the tip bias gives a direct measurement of the surface potential. In the EFM-phase method, however, the force gradient rather than the force itself is measured. This provides improved spatial resolution compared with the conventional EFM.10 In addition, the cantilever’s phase response is more sensitive than its amplitude response to changes in the tip–sample interactions and is less susceptible to height variations on the sample surface.

To prepare the sample, P3HT from Sigma-Aldrich was dissolved in CCl\(_4\) and the solution was spin coated onto SiO\(_2\)/Si surfaces to form a network structure. \( \lambda \)-DNA from

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New England Biolabs was first diluted in a MgCl\textsubscript{2} buffer, then a drop of this solution was dropped on the SiO\textsubscript{2}/Si surfaces, rinsed with flowing pure water to align the DNA molecules, and dried in flowing compressed nitrogen. After P3HT or DNA deposition, two 15-nm-thick Au electrodes were evaporated onto the sample by a wire shadow mask method.\cite{footnote1} The wire shadow left a gap of about 4 \mu m on the Au film. The width of the electrodes is about 3 mm. When making Au contact by thermal evaporation, care was taken so that organic molecules were kept at a low enough temperature to prevent decomposition, by using a shield to restrict the radiation incident of the sample, and by employing a low gold evaporation rate of 0.1 nm/s.

Figure 1(a) shows an AFM image of a network structure of P3HT formed on a SiO\textsubscript{2}/Si surface. An n\textsuperscript{+}-Si AFM tip with 300-kHz resonant frequency was employed. Its free tapping amplitude was about 13 nm, and the tip was engaged when its amplitude decreased to 85\% of its free amplitude. Similar cellular networks have been reported recently for spin-coated Au nanocrystals decorated with short alkanethiol chains on silicon.\cite{footnote12} The authors of this study argue that this cellular network cannot be explained based on Marangoni convection alone; instead, spinodal decomposition was found to play key role. It is noteworthy that very similar networks are observed in such different systems as decorated nanoclusters and semiconducting polymer molecules, indicating a universal driving mechanism. Under dc bias, the current passing through this network structure was quite unstable; increasing the film thickness of the P3HT polymer tends to stabilize the current, and \textit{I–V} measurement in air showed a rectifying behavior of the Au/P3HT/Au system. No current signal could be detected for the insulating substrate SiO\textsubscript{2}.

In order to confirm the conducting nature of the P3HT network and the electrostatic origin of the EFM-phase contrast, a 2-\mu m cut on the network along the gap direction was made with the tip, and then EFM-phase measurement was carried out. Figure 2 is the result obtained when applying a +5-V dc to the left electrode during the lift scan. The cut can be clearly seen in this image, and a potential drop exists across the cut due to the break of electrical connection.

By comparison, a standard surface potential measurement was also carried out on the Au/P3HT/SiO\textsubscript{2} structure, as shown in Fig. 3, obtained when applying a +5-V dc bias to the left electrode with the n\textsuperscript{+}-Si tip. In the surface potential

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**FIG. 1.** Images of topography and EFM-phase for P3HT with a gold contact at each end. (a) flattened topography image. (b) EFM/phase map when a bias of +5 V was applied to the left electrode, the phase range is 5°.

**FIG. 2.** EFM/phase image of a P3HT network in which a mechanical cut of 2 \mu m has been made with the tip in contact mode. The EFM/phase map was taken under identical conditions to Fig. 1(b) and shows clearly that the molecules immediately to the right of the cut are at a lower potential due to a longer connecting path around the cut. The phase range is 5°.

Since the phase lag decreased gradually from the left side to the right side in the gap region, the potential on the network also decreased gradually, consistent with a conductive network structure. On the other hand, for a clean SiO\textsubscript{2} surface, most of the potential is dropped near the edge of the left Au electrode.

In contrast, no phase shift was detected on Au electrodes, P3HT molecules, or SiO\textsubscript{2} substrate when the bias was 0 V. This means that the work function difference between these materials and the tip is not large enough to induce a phase shift on the cantilever. In the EFM-phase image, the darker region has a bigger phase lag than the brighter region due to the attractive electrostatic force on the tip, as can be seen in the two Au electrodes on the left and right sides. In the gap region, the P3HT network structure gave a clear phase shift image which coincides with its topographic image, showing a resolution better than 20 nm.
image, the brighter region has a higher potential than the darker region, as can be seen in the two Au electrodes on the left and right sides. In the gap region, the potential decreases gradually from the left side to the right side. Step structures can be seen in the surface potential image, similar to the EFM-phase image in Fig. 1(b). However, the molecular network structure did not appear in the surface potential image. It is clear from this comparison that surface potential measurement has lower lateral resolution than the EFM/phase measurement.

Figure 4(a) is an AFM image of an Au/DNA/SiO$_2$ preparation. λ-DNA strands have been stretched by flowing water and are long enough to bridge the two Au contacts. No trace of DNA strands can be found, except a few faint traces near the left Au film edge. The phase range is 5°.

![Images of topography and EFM-phase for λ-DNA. (a) Flattened topography image. (b) EFM/phase map when a bias of +5 V was applied to the left electrode. No trace of DNA strands can be found, except a few faint traces near the left Au film edge. The phase range is 5°.](image)

In conclusion, EFM has been applied to characterize the electrical behavior of P3HT and DNA molecular networks. The EFM-phase provides a surface potential map of macromolecules in a noncontact mode, with a lateral resolution of <20 nm, showing the macromolecular conductivity in a relatively straightforward manner. Quantitative analysis is complicated by the actual tip shape and surface topography and is currently being developed. Whereas networks of the semiconducting polymer P3HT are conducting, DNA strands are shown to be fairly highly insulating, with resistivity $\gg 1 \times 10^{7} \Omega$ cm. Such scanning-probe-microscopy-based techniques are likely to be crucial for characterizing, modifying, and controlling molecular networks and devices in future nanoelectronics.

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