Short Communication

Diketo acids inhibit the cap-snatching endonuclease of several *Bunyavirales*

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**ABSTRACT**

Several fatal bunyavirus infections lack specific treatment. Here, we show that diketo acids engage a panel of bunyavirus cap-snatching endonucleases, inhibit their catalytic activity and reduce viral replication of a taxonomic representative *in vitro*. Specifically, the non-salt form of L-742,001 and its derivatives exhibited EC₅₀ values between 5.6 and 6.9 μM against a recombinant BUNV-mCherry virus. Structural analysis and molecular docking simulations identified traits of both the class of chemical entities and the viral target that could help the design of novel, more potent molecules for the development of pan-bunyavirus antivirals.

The order *Bunyavirales* contains potentially fatal viruses that lack effective medical countermeasures. Many of them pose major public health threats and are pathogens prioritized in the World Health Organization (WHO) Blueprint for research and development (R&D) (Mehand et al., 2018). A hallmark is their segmented single-stranded RNA (ssRNA) genome with either negative or ambisense polarity. Thus, the synthesis of viral mRNA can be considered a limiting step in the early stages of the life cycle. The presence of 5′-capped extensions derived from the host mRNA in the viral transcripts suggests the use of a cap-snatching mechanism to initiate transcription. This mechanism was initially described for the *Influenza virus* (IFV) and although the specifics might vary between viruses, a constant feature is a metal dependent cap-snatching endonuclease (Cap-ENDO) responsible for the mRNA cleavage (Dias et al., 2009; Plicht et al., 1981; Reguera et al., 2016). Such viral enzymatic domain is essential for virus replication and presents a structurally conserved active site that can be exploited to generate broad-spectrum antivirals (Fernandez-Garcia et al., 2016; Reguera et al., 2010; Saez-Ayala et al., 2018; Song et al., 2016; Wang et al., 2020).

Very few data are available about the activity of bunyaviral Cap-ENDO inhibitors, but structural studies of the target support the idea that chelating agents as the class of diketo acids (DKAs) could be addressed as starting points for the creation of drug-like candidates (Jones et al., 2019; Reguera et al., 2010; Saez-Ayala et al., 2018). In this study, we assessed the inhibitory scope of a series of DKA compounds (Fig. 1) against Cap-ENDOs from several bunyaviruses. These small-molecules have proven to be effective inhibitors of metalloenzymes like the *Human Immunodeficiency virus* 1 integrase (HIV-1 IN) and the IFV Cap-ENDO (Bacchi et al., 2011; Sechi et al., 2004; Stevaert et al., 2015), exerting their action through coordination to the bivalent metal ion(s) in the catalytic core of the enzymes. Particularly compound (8), also known as L-742,001, has demonstrated potent antiviral activity against IFV both *in vitro* and *in vivo* (Hastings et al., 1996; Nakazawa et al., 2008; Stevaert et al., 2015). It targets the two metal cofactors of the IFV Cap-ENDO via the DKA moiety while engaging crucial amino acids with its two aromatic “wings” (DuBois et al., 2012). Interestingly enough, although site-directed mutagenesis of residues associated to the active site of IFV Cap-ENDO leads to L-742,001 resistance, a single naturally emerging variant less susceptible to L-742,001 has been reported after serial passage of IFV in the presence of this molecule...
The synthetic routes and the chemical characterization are described in the supplementary data. To determine if the compounds were able to bind to the homologous enzymes, we measured the thermal stability of LACV_{1-184} WT, RVFV_{1-248} WT and ANDV_{1-200} N167A in the presence of 25 μM of the corresponding DKA either with 10 mM EDTA or 10 mM MnCl₂. Only in the presence of the divalent cations, the compounds were able to contribute to the stability of the proteins, which is common for chelators that bind mainly through coordination to the metal cofactors. The highest increment to the melting temperature (Tm) in the presence of Mn²⁺ was observed with the addition of the prototype compound (8) and its salt L-742,001 hydrochloride (HCl), which raised the Tm (ΔTm) by more than 5 °C in all Cap-ENODs (Table 1; Fig. 2A).

Fig. 1. Schematic representation of compounds (1)–(13); in bold the chelating region.

(DKAs (8), (10), and (12) showed increments of the Tm bigger than the corresponding esters (9), (11), and (13), in accord with the higher complexing ability towards metal ions of the carboxyl group when compared to the ester group (Supplementary Fig. S1). Although the thermal shift assay (TSA) results might be in agreement with the ligands association constants, it can be misleading to rank compounds by their potency using the ΔTm values (Lo et al., 2004; Pantoliano et al., 2001).

Therefore, to evaluate the inhibitory capacity of DKAs, we further investigated the activity of all those that undoubtedly engaged the enzymes and exhibited a ΔTm ≥ 3 °C.

A FRET-based nuclease-monitoring assay was employed to determine the half maximal inhibitory concentration (IC₅₀) of selected compounds. Serial dilutions of the respective compounds were pre-incubated for 15 min with 0.25 μM of the proteins in a mixture consisting of 50 mM Tris-HCl pH 7.5, 50 mM NaCl, 5% glycerol, 0.05 U/μl RNasin (Promega) and 10 mM MnCl₂ or EDTA. The reactions were initiated by the addition of 1 μM of a 12mer poly(A) ssRNA substrate labeled at the 5’ end with fluorophore 6-FAM and the 3’ end with quencher BHQ-1 purchased from Biomers. The fluorescence signal was measured every 10 s for 20 min in a 96-well plate format at wavelengths 490 nm and 515 nm for excitation and emission, respectively. The initial velocity of the reactions (V₀) was determined as the slope of the linear phase of the progress curves, and the percentages of the enzymatic activity were plotted against the concentrations of the compounds on a semilogarithmic graph. The IC₅₀ of the compounds was calculated by nonlinear regression analysis performed using GraphPad Prism version 7.00 (GraphPad Software, La Jolla California USA, www.graphpad).

Table 1

<table>
<thead>
<tr>
<th>Compound</th>
<th>ΔTm (°C)</th>
<th>IC₅₀ (μM)</th>
<th></th>
<th></th>
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<tbody>
<tr>
<td></td>
<td>LACV_{1-184} WT</td>
<td>ANDV_{1-200} N167A</td>
<td>RVFV_{1-248} WT</td>
<td>LACV_{1-184} WT</td>
</tr>
<tr>
<td>(1)</td>
<td>3.5 ± 0.5</td>
<td>4.9 ± 0.1</td>
<td>5.6 ± 0.2</td>
<td>24.9 ± 2.6</td>
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<tr>
<td>(2)</td>
<td>4.9 ± 0.3</td>
<td>3.5 ± 0.5</td>
<td>4.9 ± 0.03</td>
<td>NT</td>
</tr>
<tr>
<td>(3)</td>
<td>1.4 ± 0.01</td>
<td>1.4 ± 0.1</td>
<td>2.2 ± 0.1</td>
<td>NT</td>
</tr>
<tr>
<td>(4)</td>
<td>2.2 ± 0.8</td>
<td>2.9 ± 0.1</td>
<td>2.5 ± 1</td>
<td>NT</td>
</tr>
<tr>
<td>(5)</td>
<td>2.1 ± 0.2</td>
<td>1.9 ± 0.2</td>
<td>2.9 ± 0.1</td>
<td>NT</td>
</tr>
<tr>
<td>(6)</td>
<td>3 ± 0.3</td>
<td>2.1 ± 0.1</td>
<td>2.2 ± 0.2</td>
<td>NT</td>
</tr>
<tr>
<td>(7)</td>
<td>4.4 ± 0.1</td>
<td>3.2 ± 0.4</td>
<td>4.4 ± 0.3</td>
<td>31.0 ± 3.1</td>
</tr>
<tr>
<td>L-742,001 HCl</td>
<td>6.3 ± 0.3</td>
<td>6.4 ± 1</td>
<td>6.1 ± 0.2</td>
<td>4.0 ± 0.5</td>
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<tr>
<td>(8)</td>
<td>6.9 ± 0.9</td>
<td>6.1 ± 0.1</td>
<td>5.5 ± 0.3</td>
<td>7.3 ± 0.8</td>
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<tr>
<td>(9)</td>
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<td>1.7 ± 0.3</td>
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<tr>
<td>(10)</td>
<td>5 ± 0.8</td>
<td>4.1 ± 0.2</td>
<td>3.7 ± 0.6</td>
<td>13.7 ± 2.3</td>
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<tr>
<td>(11)</td>
<td>3.8 ± 0.5</td>
<td>2.6 ± 0.3</td>
<td>1.6 ± 0.3</td>
<td>NT</td>
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<tr>
<td>(12)</td>
<td>3.7 ± 0.1</td>
<td>4.3 ± 0.01</td>
<td>3.7 ± 0.1</td>
<td>11.4 ± 1.9</td>
</tr>
<tr>
<td>(13)</td>
<td>1.2 ± 0.8</td>
<td>2.8 ± 0.1</td>
<td>2.5 ± 0.3</td>
<td>NT</td>
</tr>
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</table>

ΔTm- The compound contribution to the enzyme thermal stability was determined by the difference in Tm between the samples containing MnCl₂ in the presence and absence of the respective compound; IC₅₀- The half-maximal inhibitory concentration was calculated by fitting the dose-response curves using a four parameter nonlinear regression; NT- Not tested. All data are mean ± standard deviation from 3 independent experiments.
(1) and its ester (2) as well as DKA (7) were able to suppress RVFV, ANDV and, to a lesser extent, also LACV Cap-ENDOs. Compounds (8) and its salt L-742,001 HCl, (10) and (12) were the most effective inhibitors of the RNA cleavage, displaying IC$_{50}$ values within the lower μM range towards all the Cap-ENDOs tested (Table 1). As shown for compound (8), all the DKAs tested had a similar dose-dependent effect over the catalytic activity of the different Cap-ENDOs (Fig. 2B).

In agreement with our results, L-742,001 has also been reported to inhibit the Cap-ENDO of the Lymphocytic Choriomeningitis virus (LCMV) (Saez-Ayala et al., 2018). The enzymatic domain of this other bunyavirus belongs to the subgroup of His$^+$ Cap-ENDOs, which characteristically possess a substantially reduced activity and are suggested to be regulated differently as the His$^+$ Cap-ENDOs used in this study (Holm et al., 2018). Thus, to investigate the binding of L-742,001 to the different Cap-ENDOs, we performed a series of molecular docking experiments, using the available structures for LACV (PDB: 2XI7), ANDV (PDB: 6Q99), and RVFV (homology model). All results are shown from the same orientation centered on the Mn$^{2+}$ ions (purple), and display the similarities in the predicted binding between the different Cap-ENDOs. The hydrophobic contact area of a defined sub-pocket is represented in green.

![Fig. 2. L-742,001 inhibitory molecular interactions with Bunyavirales Cap-ENDO’s.](image-url)

**Fig. 2.** L-742,001 inhibitory molecular interactions with Bunyavirales Cap-ENDO’s. (A) Thermal stability of LACV L$_{1-184}$ WT (left panel), ANDV L$_{1-200}$ N167A (middle panel) and RVFV L$_{1-248}$ WT (right panel) in the presence of 10 mM EDTA (black dashed line), 10 mM EDTA plus 25 μM L-742,001 (green dashed line), 10 mM MnCl$_2$ (orange solid line) and 10 mM MnCl$_2$ plus 25 μM L-742,001 (green solid line). The compound contribution to the stability of the proteins was determined by the difference in Tm between the samples containing MnCl$_2$ in the presence and absence of L-742,001 (ΔTm). The lines depict the non-linear fits of the normalized fluorescence curves to the Boltzmann equation. (B) Dose-dependent inhibition of the nuclease activity of LACV L$_{1-184}$ WT (red), ANDV L$_{1-200}$ N167A (blue) and RVFV L$_{1-248}$ WT (grey) by L-742,001. The reactions were initiated with the addition of 1 μM of FRET-labeled ssRNA substrate to the pre-incubated mixtures of 0.25 μM of the respective enzymes with two-fold serial dilutions of L-742,001. The fluorescent signal arising from the substrate cleavage was measured using a Roche LightCycler 480 II at λ$_{ex}$/λ$_{em}$ = 490 nm/515 nm. The V$_0$ of the reactions were used to calculate the percentage of activity at the different concentrations of the inhibitor. The dose-response curves were fit using a four parameter nonlinear regression. ΔTm and half maximal inhibitory concentration (IC$_{50}$) values are the mean and standard deviation from 3 independent experiments. Final 5% DMSO concentration was maintained in all assays to favor compound solubility. (C) Molecular docking results for L-742,001 on the Cap-ENDOs of LACV (PDB: 2XI7), ANDV (PDB: 6Q99), and RVFV (homology model). All results are shown from the same orientation centered on the Mn$^{2+}$ ions (purple), and display the similarities in the predicted binding between the different Cap-ENDOs. The hydrophobic contact area of a defined sub-pocket is represented in green.
001 complex; a common hydrophobic sub-pocket can be identified in the binding pocket, occupied by the chlorobenzyl ring of L-742,001 in all results obtained from the molecular docking as well as in the complex structure with LCMV domain (Supplementary Fig. S2); the tertiary amine of L-742,001 establishes a clear interaction respectively with Asp45 of the LACV Cap-ENDO and with Glu54 of the ANDV Cap-ENDO, while sits in the proximity of Asp62 in the RVFV Cap-ENDO model – all these three acidic residues are also involved in the coordination of an Mn$^{2+}$ ion. These observations could potentially be exploited in the design of a novel series of more potent, broad-spectrum, Cap-ENDO inhibitors.

Having confirmed the inhibitory effect of L-742,001 and its salt L-742,001 HCl against a diverse set of bunyaviral Cap-ENDOs in enzymatic assays, we studied its ability to reduce viral replication in a cell-based assay. The antiviral activity against Bunyamwera virus (BUNV)-mCherry of L-742,001 HCl, L-742,001 and derivatives were evaluated in A549 cells (Shi et al., 2010). BUNV is considered a prototypic bunyavirus and is used as a model for research of pathogens within Bunyavirales. Serial dilutions of compounds were added followed by infection with 500 CCID$_{50}$ of virus. At 4 days post infection (dpi) cells were stained with a 2.5 μg/ml Hoechst solution and high content imaging analysis was performed on the Arrayscan XTI from Thermofisher® using a custom protocol based on the SpotDetector BioApplication (Cellomics® software). The cell nuclei were detected using a dynamic threshold method based on differences in intensity peaks and the mCherry signal spots were detected by using a fixed intensity threshold based on the images acquired from the mCherry positive control (Fig. 3B). EC$_{50}$ values are defined as the compound concentration that reduces the number of mCherry signal spots by 50% compared to the virus control (Fig. 3C). CC$_{50}$ values are defined as the compound concentration that reduces the number of cells based on the nuclei by 50% compared to the virus control. In addition, we assessed the effect of the compounds on virus yield by RT-qPCR. Per reaction, a 20 μl RT-qPCR mix was prepared containing 4 μl of viral RNA with 600 nM and 900 nM of FW and REV primers, respectively (FW: 5'-ACACCACTGGGCTTAG-3', REV: 5'-CAGCCCCCAAGGTAA-3'). For absolute quantification, standard curves were generated using 10-fold dilutions of template DNA of known concentration.

L-742,001 HCl inhibited BUNV-mCherry in a dose-dependent manner with EC$_{50}$ values of $10.6 \pm 1.6 \, \mu M$ and $18.5 \pm 1.6 \, \mu M$, as determined by high content imaging and RT-qPCR, respectively, with no cytotoxicity for the host cell (CC$_{50}$ > 100 μM) (Fig. 3A). A similar profile was observed with the derivatives of L-742,001, both esters and carboxylic acids. Their EC$_{50}$ is slightly lower than L-742,001 HCl, although comparable to that of the parent non-salt form (Table 2). Unexpectedly, albeit presenting equivalent $^1$H-NMR spectra (Supplementary Fig. S3), we observed higher cytotoxicity for the non-salt form of L-742,001 HCl.

![Fig. 3. L-742,001 in vitro antiviral activity against Bunyaviruses.](image-url)

(A) Effect of L-742,001 HCl on BUNV-mCherry replication in A549 cells. Viral mCherry signal spots (black bars) and cell viability (solid circles) were determined using high-content imaging analysis ($n = 7$). Viral RNA yield (white bars) were determined by RT-qPCR ($n = 5$). All data are mean ± standard deviation. (B) Representative images acquired through high content imaging of virus control (VC), cell control (CC). (C) Representative images acquired through high content imaging of BUNV-mCherry infected cells treated with a serial dilution of L-742,001 HCl. Images represent one out of nine fields acquired per well.
In summary, our results demonstrate that the Cap-ENDO of bunyaviruses is a valid target for the development of broad-spectrum antivirals and that the DKAs class of compounds contains chemical features worth exploring in the design of novel inhibitors. We have documented the biochemical inhibition of several bunyaviral Cap-ENDOs by different DKAs and have confirmed that for a group of them, this translates to a modest antiviral activity as observed in the reduction of the number of mCherry spots and the viral RNA yield of the recombinant BUNV-mCherry virus. Moreover, our molecular docking simulations have provided us with some valuable information on the interaction of L-742,001 with this viral enzyme, defining a possible basic common pharmacophore for a broad-spectrum class of Cap-ENDO inhibitors. From these results, more potent and selective molecules with a wide range of action can now be designed. Hence, bringing us one step closer to a pan-bunyavirus antiviral.

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References